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- (71) Applicant (*for all designated States except US*): E.I. DU PONT DE NEMOURS AND COMPANY [US/US]; 1007 Market Street, Wilmington, DE 19898 (US). (88) Date of publication of the international search report: 1 July 2004
- (72) Inventor; and
- (75) Inventor/Applicant (*for US only*): DEAN, David, M. [US/US]; 1073 Forest Road, West Chester, PA 19382 (US). *For two-letter codes and other abbreviations, refer to the "Guidance Notes on Codes and Abbreviations" appearing at the beginning of each regular issue of the PCT Gazette.*

(54) Title: ETHYLENE COPOLYMER WITH FLUORINE-CONTAINING GRAFT AND A PROCESS FOR PREPARING THE SAME

(57) Abstract: The present invention is fluorine-containing ethylene copolymer composition and a process of preparing same. The ethylene copolymers of the present invention are melt processible fluorinated copolymers having surface energies of less than about 25 dynes/cm.

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TITLE**ETHYLENE COPOLYMER WITH FLUORINE-CONTAINING GRAFT
AND A PROCESS FOR PREPARING SAME**

5 This application claims the benefit of U.S. Provisional Application
No. 60/421,946, filed October 29, 2002.

BACKGROUND OF THE INVENTION**Field of the Invention**

10 The present invention relates to fluorine-containing ethylene
copolymer compositions having grafted fluorine-containing functionality.
The present invention also relates to polymer blends obtained from
fluorine-containing ethylene copolymers described herein. The present
invention also relates to a process for preparing the fluorine-containing
15 ethylene copolymers of the present invention.

Discussion of the Related Art

Fluoropolymer compositions are widely used for surface
modification, as agents to impart desirable surface properties to various
types of surfaces. For example, fluoropolymer compositions can impart or
20 enhance the water and oil repellency of certain surfaces, including fabrics
and upholstery. However, surface treatment using fluoropolymers can
involve complex processing steps to ensure that the fluoropolymer is
applied and bonded to the surface being treated. The process can be
difficult and expensive. Organic solvent vapors can be released to the
25 atmosphere during the processing. Surface treatment can involve high
temperature curing of the fluoropolymer to the substrate surface.

To impart water and oil repellency, fluorochemicals or
fluoropolymers can be dissolved or dispersed either in organic solvents or
in water. For example, mixtures of fluorinated copolymers, mainly
30 comprising perfluoroalkyl methacrylate, and vinyl copolymers are
disclosed in U.S. Pat. No. 3,277,039. U.S. Pat. No. 2,803,615 discloses
acrylate/methacrylate esters of N-alkyl or N-alkanol
perfluoroalkanesulfonamides used to impart grease and oil repellency.

Fluorochemical compositions for treating textile fibers and fabrics comprising an aqueous solution or dispersion of a fluorochemical acrylate and a polyalkoxylated polyurethane having pendant perfluoroalkyl groups is described in U.S. Pat. No. 5,350,557. U.S. Pat. No. 5,536,304
5 describes a composition for imparting water and oil repellency comprising a fluoroaliphatic radical containing agent, and a cyclic carboxylic acid anhydride-containing polysiloxane.

Fluorochemicals can be melt-blended with thermoplastic polymers, and thereby impart water and oil repellency to the polymer by migrating to
10 the polymer surface as described in, for example, U.S. Pat. No. 5,025,052, wherein the preparation of fluoroaliphatic radical-containing oxazolidinone compositions for blending with thermoplastic polymers is described. U.S. Pat. No. 5,380,778 describes thermoplastic compositions comprising fluoroaliphatic radical containing aminoalcohols and a thermoplastic
15 synthetic organic polymer.

Ethylene copolymers are useful polymeric materials in many applications. Ethylene copolymers can find use in applications such as packaging, laminate films, and adhesives for example. Conventional polyolefins such as polypropylene, polyethylene, and conventional
20 ethylene copolymers have high surface tension relative to fluoropolymers such as polytetrafluoroethylene, for example. As a result, for applications wherein water and oil repellency is important, articles made from polyolefins must be treated to attain a satisfactory level of repellency. However, due to their relatively low melting point and lack of reactive
25 functional groups, treatment of polyolefins with fluorochemicals or fluoropolymers is, in general, much more difficult than treating other thermoplastic polymers.

Copolymers of ethylene and fluorine-containing monomers are known. For example, Tefzel®, manufactured by E.I DuPont de Nemours
30 and Company, is a copolymer of ethylene and tetrafluoroethylene. Copolymers of this type are very different from conventional polyethylene copolymers in many aspects. For example, fluorine-containing ethylene

copolymers are typically melt-processable only at much higher temperature than conventional ethylene polymers and ethylene copolymers, and the properties of fluorine-containing ethylene copolymers differ from conventional ethylene copolymers. Copolymers of this type are
5 not amenable to manufacture or processing under the type of conditions used to manufacture and process conventional ethylene copolymers. Ethylene/fluoromonomer copolymers of this type are not compatible in systems that currently use polyethylene copolymers. For example, known conventional ethylene/tetrafluoroethylene copolymers have no adhesion to
10 polyethylene.

It would be desirable to have a fluorine-containing ethylene copolymer that can be processed in the same way as conventional ethylene copolymers. It would be desirable to have a melt-processable fluorine-containing ethylene copolymer. It would be desirable to have a
15 fluorine-containing ethylene copolymer that has a low surface tension. It would be desirable to have a fluorine-containing ethylene copolymer that can be compatible with, or used in place of conventional ethylene copolymers.

SUMMARY OF THE INVENTION

20 In one aspect, the present invention is a fluorine-containing ethylene copolymer composition comprising: the product of the reaction between an ethylene/glycidyl (meth)acrylate copolymer and a fluorine-containing carboxylic acid.

In still another aspect, the present invention is a blend comprising
25 at least two thermoplastic materials wherein at least one is a fluorine-containing ethylene copolymer composition comprising the product of the reaction between an ethylene/glycidyl (meth)acrylate copolymer and a fluorine-containing carboxylic acid.

In another aspect, the present invention is an article having a
30 surface with a total surface energy of less than 25 dyne/cm comprising: a fluorine-containing ethylene copolymer composition comprising the

product of the reaction between an ethylene/glycidyl (meth)acrylate copolymer and a fluorine-containing carboxylic acid.

In another aspect, the present invention is a stain-resistant fiber comprising a fluorine-containing ethylene copolymer composition
5 comprising the product of the reaction between an ethylene/glycidyl (meth)acrylate copolymer and a fluorine-containing carboxylic acid.

In still another aspect, the present invention is an article formed by injection molding or by extrusion comprising a fluorine-containing ethylene copolymer composition comprising the product of the reaction between an
10 ethylene/glycidyl (meth)acrylate copolymer and a fluorine-containing carboxylic acid.

In still another aspect the present invention is a mold release additive comprising a fluorine-containing ethylene copolymer composition comprising the product of the reaction between an ethylene/glycidyl
15 (meth)acrylate copolymer and a fluorine-containing carboxylic acid.

DETAILED DESCRIPTION OF THE INVENTION

In one embodiment, the present invention is a composition comprising a fluorine-containing ethylene copolymer. The fluorine-containing ethylene copolymer of the present invention can be obtained,
20 for example, by the process of reactive melt compounding a glycidyl methacrylate containing ethylene copolymer with a fluorine-containing material that also includes carboxylic acid functionality. Reactive melt compounding, as the term is used herein, includes any process whereby reactive components are combined and reacted at temperatures above
25 the melting point of all of the reactive components. Reactive melt compounding does not require the presence of a solvent.

Suitable EGMA copolymers are copolymers obtained by copolymerization of glycidyl methacrylate with ethylene. EGMA copolymers suitable for the present invention have a weight average
30 molecular weight (M_w) of at least 50,000. Preferably the M_w of EGMA copolymers suitable for use herein is in the range of from about 60,000 to about 200,000, more preferably in the range of from about 70,000 to about

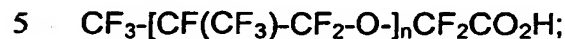
150,000 , and most preferably in the range of from about 80,000 to about 120,000. Suitable ethylene/glycidyl methacrylate copolymers (EGMA copolymers) can be obtained commercially from, for example, Aldrich Chemical Co.

5 EGMA copolymers suitable for use herein comprise from about 0.5 wt% GMA to about 30 wt%. Preferably EGMA copolymers comprise from about 2 wt% to about 18 wt% GMA, and more preferably from about 5 wt% to about 15 wt%. Most preferably EGMA copolymers comprise from about 6 wt% to about 12 wt% GMA.

10 Also EGMA copolymers suitable for use herein are terpolymers comprising ethylene, GMA and a third comonomer. Suitable third comonomers are selected from unsaturated carboxylic esters having from 3 to 12 carbons. Suitable third comonomers include, for example: n-butyl acrylate (nBA); methyl (meth)acrylate (M(M)A); ethyl (meth)acrylate; iso-
15 butyl acrylate (iBA); and vinyl acetate (VA). Methacrylate esters, for the purposes of the present invention, are suitable equivalents for acrylate esters for all purposes of the present invention. In the present invention a shorthand notation for "methacrylate and/or acrylate" can be either (meth)acrylate or, alternatively, (M)A. For example, in the present
20 application, a shorthand notation for "methyl methacrylate and/or methyl acrylate" can be either "methyl (meth)acrylate" or, alternatively, "M(M)A".

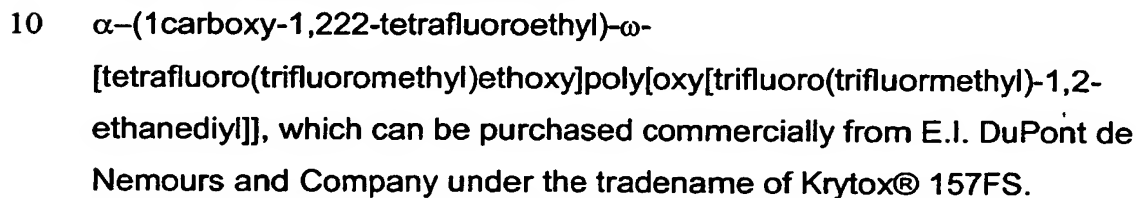
 EGMA copolymers are reacted with fluorinated compounds having carboxylic acid functionality to yield the fluorine-containing ethylene copolymers of the present invention. Suitable fluorine containing
25 carboxylic acids are any such acids that can, after reaction with an EGMA copolymer or terpolymer, yield copolymers having surface energies of preferably less than about 25 dyne/cm. Perfluorocarboxylic acids and perfluoropolyether carboxylic acids are preferred in the practice of the present invention. Suitable perfluorinated carboxylic acids can have from
30 2 to 30 carbon atoms, preferably from 4 to 24 carbons, more preferably from 6 to 20 carbons, and most preferably from 6 to 18 carbons. Suitable acids include, for example: perfluoro nonanoic acid (PFNON); perfluoro

octanoic acid (PFOCT); perfluoro heptanoic acid (PFHEP); perfluoro hexanoic acid (PFHEX). Perfluoropolyether carboxylic acids (PFPE) suitable for use in the practice of the present invention include PFPE of the general formula:



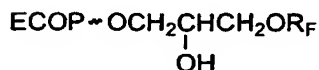
wherein n is an integer in the range of from about 5 to about 50.

Preferably n is in the range of from about 12 to about 26, more preferably in the range of from about 14 to about 24, and most preferably in the range of from about 16 to about 22. A suitable PFPE, for example, is

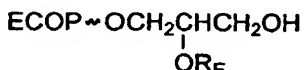


EGMA copolymers can be reacted with suitable fluorine-
 15 containing carboxylic acids by reacting the components in the melt, that is, at a temperature above the melting range of both reactants. Melt blending or coextrusion of the components results in the acid-catalyzed ring opening of the three member heterocyclic ring (epoxide) of the GMA moiety to yield a perfluoroester of glycerol grafted to the ethylene
 20 copolymer (grafted fluoroester copolymer, or GFEC). A GFEC of the present invention can have a general formula as defined in any of Structures 1 –3 below.

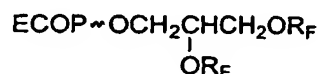
Structure 1:



Structure 2:



Structure 3:



wherein:

ECOP is the ethylene copolymer backbone;

R_F is a fluorine-containing acyl group suitable for use in the practice of the present invention.

25 A fluorine-containing ethylene copolymer product of the reaction as described herein can be a complex mixture of any of the above structures,

in addition to other reaction products obtained according to the process described herein, in addition to unreacted starting materials. A composition of the present invention comprises GFEC having any of Structures 1, 2 or 3 either individually present or present in any
5 combination, in an amount of at least about 15 wt% of the total composition. Preferably, the GFEC comprises at least about 20 wt% of the total composition. More preferably the GFEC comprises at least about 25 wt%, and most preferably at least about 30 wt% of the composition. Of the GFEC present, preferably Structure 3 comprises less than 10 wt% of
10 the combined total weight of the GFEC, more preferably less than 5 wt%, and most preferably less than 2 wt%.

In another embodiment, the present invention is a process for preparing a GFEC of the present invention. As noted hereinabove, a GFEC of the present invention can be obtained by reactive melt
15 compounding a suitable glycidyl methacrylate-containing ethylene copolymer and a suitable fluorine-containing carboxylic acid. For a melt compounding process according to the present invention, the components are preferably combined at a temperature above about 160°C but less than about 250°C. The components can be blended in the melt using
20 conventional mixing means, such as a Haake mixer, for example. The components can also be coextruded in the melt (reactive coextrusion) to obtain the GFEC product of the present invention.

EXAMPLES

The Examples and the Comparative Examples described herein are for illustrative purposes only, and not meant to limit the scope of the present invention in any manner.

Example 1

A mixture of 80.3 wt% EGMA copolymer (8 wt% GMA) and 19.7 wt% PFNON was heated to a temperature of 180°C and blended in a Haake mixer for 10 minutes. The resulting material was then pressed between two Teflon® FEP plates at 180°C for 10 minutes to form 10 mil thick films which were then cut into pieces. The film pieces were washed in an acetone bath for 2 days, and then rinsed for 1 minute in acetone and dried at room temperature. The film samples were analyzed by infra red (IR) spectroscopy to show absorption in the region of 1787 cm⁻¹, indicative of the carbonyl absorption of a perfluoroester group. The surface energy was analyzed and is reported in Table 1.

Examples 2-6

The process of Example 1 was repeated using the components in Table 1.

Comparative Examples 1-3

The procedure of Example 1 was repeated with the starting polymers indicated in Table 1, without added grafting agent.

Table 1 Surface Energy Measurements of 10 mil Compression Molded Film

Example	Starting Polymer	Grafting Agent	Total Surface Energy (dyne/cm)
Comparative 1	Teflon® FEP	None	19.1
Comparative 2	DPE 2020	None	33.8
Comparative 3	E/ 8 wt% GMA	None	27.6
Ex 1	E/ 8 wt% GMA	19.7 wt% PFNON	19.2
Ex 2	E/ 8 wt% GMA	5.9 wt% PFPE	20.8
Ex 3	E/ 8 wt% GMA	11.1 wt% PFPE	16.6
Ex 4	E/ 8 wt% GMA	17.9 wt% PFOCT	20.8
Ex 5	E/ 8 wt% GMA	16.1 wt% PFHEP	23.4
Ex 6	E/ 8 wt% GMA	14.2 wt% PFHEX	20.5

CLAIMS**WHAT IS CLAIMED IS:**

1. A fluorine-containing ethylene copolymer composition comprising:
the product of the reaction between an ethylene/glycidyl
5 (meth)acrylate copolymer and a fluorine-containing carboxylic acid.
2. A fluorine-containing ethylene copolymer composition comprising:
the product of the reaction between an ethylene/glycidyl
(meth)acrylate copolymer and a perfluorinated carboxylic acid,
characterized in that the copolymer absorbs light in the region of
10 from about 1750 cm^{-1} to about 1800 cm^{-1} of the infra red absorption spectrum.
3. A blend comprising at least two thermoplastic materials wherein at
least one is a fluorine-containing ethylene copolymer composition
comprising the product of the reaction between an ethylene/glycidyl
15 (meth)acrylate copolymer and a fluorine-containing carboxylic acid.
4. A blend comprising at least two thermoplastic materials wherein at
least one is a fluorine-containing ethylene copolymer composition
comprising the product of the reaction between an ethylene/glycidyl
(meth)acrylate copolymer and a fluorine-containing carboxylic acid,
20 characterized in that the copolymer absorbs light in the region of
from about 1750 cm^{-1} to about 1800 cm^{-1} of the infra red absorption spectrum.
5. An article having a surface with a total surface energy of less than
25 dyne/cm comprising: a fluorine-containing ethylene copolymer
composition comprising the product of the reaction between an
ethylene/glycidyl (meth)acrylate copolymer and a fluorine-
containing carboxylic acid.
6. An article having a surface with a total surface energy of less than
25 dyne/cm comprising: a fluorine-containing ethylene copolymer
30 composition comprising the product of the reaction between an
ethylene/glycidyl (meth)acrylate copolymer and a fluorine-
containing carboxylic acid, characterized in that the copolymer
absorbs light in the region of from about 1750 cm^{-1} to about 1800 cm^{-1} of the infra red absorption spectrum.
- 35 7. A stain-resistant fiber comprising a fluorine-containing ethylene
copolymer composition comprising the product of the reaction

between an ethylene/glycidyl (meth)acrylate copolymer and a fluorine-containing carboxylic acid.

8. A stain-resistant fiber comprising a fluorine-containing ethylene copolymer composition comprising the product of the reaction between an ethylene/glycidyl (meth)acrylate copolymer and a fluorine-containing carboxylic acid, characterized in that the copolymer absorbs light in the region of from about 1750 cm^{-1} to about 1800 cm^{-1} of the infra red absorption spectrum.
9. An article formed by injection molding or by extrusion comprising a fluorine-containing ethylene copolymer composition comprising the product of the reaction between an ethylene/glycidyl (meth)acrylate copolymer and a fluorine-containing carboxylic acid.
10. An article formed by injection molding or by extrusion comprising a fluorine-containing ethylene copolymer composition comprising the product of the reaction between an ethylene/glycidyl (meth)acrylate copolymer and a fluorine-containing carboxylic acid, characterized in that the copolymer absorbs light in the region of from about 1750 cm^{-1} to about 1800 cm^{-1} of the infra red absorption spectrum.
11. A mold release additive comprising a fluorine-containing ethylene copolymer composition comprising the product of the reaction between an ethylene/glycidyl (meth)acrylate copolymer and a fluorine-containing carboxylic acid.
12. A mold release additive comprising a fluorine-containing ethylene copolymer composition comprising the product of the reaction between an ethylene/glycidyl (meth)acrylate copolymer and a fluorine-containing carboxylic acid, characterized in that the copolymer absorbs light in the region of from about 1750 cm^{-1} to about 1800 cm^{-1} of the infra red absorption spectrum.

INTERNATIONAL SEARCH REPORT

International Application No

PCT/US 03/34735

A. CLASSIFICATION OF SUBJECT MATTER

IPC 7 C08F8/18 C08F210/02

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 7 C08F C08K C08L

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

EP0-Internal, WPI Data

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	US 5 141 661 A (DEROSA THOMAS F ET AL) 25 August 1992 (1992-08-25) the whole document	1,2
A	WO 02/072648 A (DU PONT ;CHOU RICHARD T (US)) 19 September 2002 (2002-09-19) page 3, line 28 -page 4, line 13; claims	1-12



Further documents are listed in the continuation of box C.



Patent family members are listed in annex.

* Special categories of cited documents:

"A" document defining the general state of the art which is not considered to be of particular relevance

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Date of the actual completion of the international search

23 April 2004

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04/05/2004

Name and mailing address of the ISA

European Patent Office, P.B. 5818 Patentlaan 2
NL - 2280 HV Rijswijk
Tel. (+31-70) 340-2040, Tx. 31 651 epo nl,
Fax (+31-70) 340-3016

Authorized officer

Baekelmans, D

INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No
PCT/US 03/34735

Patent document cited in search report		Publication date	Patent family member(s)	Publication date
US 5141661	A	25-08-1992	NONE	
WO 02072648	A	19-09-2002	CA 2438317 A1	19-09-2002
			EP 1366092 A2	03-12-2003
			WO 02072648 A2	19-09-2002